The Stereocontrolled Total Synthesis of Altohyrtin A/Spongistatin 1. Part 4: Fragment Couplings, Completion of the Synthesis, Analogue Generation and **Biological Evaluation**

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General Experimental Details

¹H nuclear magnetic resonance (NMR) spectra were recorded at either 250, 400 500 or 800 MHz on Bruker DPX 250, DPX 400, DRX 500 or DRX 800 spectrometers at ambient temperature using an internal deuterium lock. The following internal references were used for the residual protons in the following solvents: CDCl₃ (δ_H 7.26), C₆D₆ (δ_H 7.16) and CD₃CN (δ_H 1.94). Data are presented as follows: chemical shift (in ppm on the δ scale relative to tetramethylsilane $\delta_{TMS} = 0$), integration, multiplicity, coupling constant and interpretation XX-CH where XX refers to the carbon no. to which the proton in question is attached. Where reasonable, this numbering is based on the spongistatin skeleton. The following abbreviations for splitting patterns are used: s, singlet; d, doublet; t, triplet; q, quartet; quin., quintet; m, multiplet; br, broad. When the multiplet is derived from couplings to non-equivalent protons with coincidentally the same coupling constants then the multiplet is referred to as app, apparent. Assignments were determined either on the basis of unambiguous chemical shift or coupling pattern, COSY experiments or by analogy to fully interpreted spectra for related compounds. ¹³C nuclear magnetic resonance (NMR) spectra were recorded at 100.6 MHz or 62.5 MHz on Bruker AM 400 or DPX 250 spectrometers respectively at ambient temperature using an internal deuterium lock, and all chemical shift values are reported in parts per million (δ) downfield relative to tetramethylsilane (TMS, $\delta_{TMS} = 0$). An internal reference was used for CDCl₃ (δ_C 77.16) and C₆D₆ (δ_C 128.06).

Infra-red spectra were recorded on Perkin-Elmer 1620 (FT-IR) spectrometers using 0.5 cm sodium chloride plates. Absorbance bands are reported in wavenumbers (cm⁻¹) relative to polystyrene as the calibrant, and the following abbreviations are used to describe their appearance: w, weak; s, strong; br, broad. Only the most significant bands are reported.

High and low resolution mass spectra were acquired using positive chemical ionisation using NH₄⁺ (+CI, NH₃) by the EPSRC National Mass Spectrometry Service Centre, Swansea, UK and the Departmental Mass Spectrometry Service, University Chemical Laboratory, Cambridge, using Electronic Supplementary Material for Organic & Biomolecular Chemistry

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electron impact (EI), electrospray (+ESI), chemical ionisation (+CI) or fast atom bombardment (+FAB) ionisation techniques. The parent ion $[M]^+$ or $[MH]^+$ or $[M + NH_4]^+$ is quoted, followed by significant fragments with their relative intensities.

Optical rotations were recorded on a Perkin Elmer 241 polarimeter at the sodium D line (589 nm) and are reported as follows: $[\alpha]_{b}^{20}$, concentration (*c* in g/100 mL) and solvent (all the rotations were measured at a temperature of 20 °C). Melting points were recorded on a Kofler hot-stage and are uncorrected.

Analytical thin layer chromatography (TLC) was carried out using Merck Kieselgel 60 F_{254} plates with visualisation either by ultra violet light (254 nm), anisaldehyde or Goofy's dips. Flash column chromatography was carried out using Merck Kieselgel 60 (230-400 mesh) under a positive pressure using distilled solvents and in this thesis the term implies subsequent removal of the solvents *in vacuo* unless otherwise stated. High Performance Liquid Chromatography (HPLC) was carried out using a Rainin Instrument Co. Inc. DYNAMAX Macro-HPLC column (internal diameter: 21.4 mm), prepacked with 8 micron irregular silica particles, and equipped with a Gilson refractive index detector (Model 131) or a Gilson UV detector (Model 111B) at a wavelength of 254 nm. A flow rate of 10 mL min⁻¹ was used and all solvents were vacuum-filtered and degassed prior to use.

Reagents and solvents were prepared using standard means.¹ Anhydrous CH₂Cl₂, MeOH and hexane were distilled from CaH₂ and stored under argon; ether was distilled from sodium metal/benzophenone ketyl and stored under an argon atmosphere; THF was distilled from either LiAlH₄ or potassium metal/benzophenone ketyl and stored under an argon atmosphere. Triethylamine (Et₃N), *i*-Pr₂NEt, pyridine and 2,6-lutidine were distilled from and stored over CaH₂. Acetic acid (AcOH) was distilled from CrO₃ and Ac₂O and stored under an argon atmosphere. Simple aldehydes were distilled from calcium chloride immediately prior to use. All other reagents were used as received except where noted in the experimental procedure.

All experiments were performed under anhydrous conditions, utilising anhydrous solvents, under an atmosphere of argon, except where stated, using oven-dried glassware and employing standard techniques in handling air-sensitive materials. All reactants added *via* cannula were added using a positive pressure of argon. Where a reaction temperature is not specified the reaction was performed at room temperature. Where a compound has been published in the literature, all spectroscopic and physical properties matched those reported.

Experimental Procedures and Product Characterisation Data

(2*R*,4*S*,5*S*,6*R*)-4-(*t*-Butyldimethylsiloxy)-6-(4-(triphenylphosphonium)-butyl)-2-methoxy-2-[[4,5-(*R*,*R*)-bis-(*p*-methoxybenzyloxy)-3-(*R*)-methyl-6-(*R*)-(2-methylallyl)-tetrahydropyran-2-(*R*)-yl]-((*S*)-*p*-methoxybenzyloxy)-methyl]-5-methyl-tetrahydropyran iodide (14)

To a solution of chloride 13^2 (29.8 mg, 0.032 mmol) in MeCN/MeOH (9:1, 0.5 mL) was added ^{*i*}Pr₂NEt (11 µL, 0.064 mmol, 2 equiv.), NaI (121 mg, 0.8 mmol, 25 equiv.) and PPh₃ (422 mg, 1.6 mmol, 50 equiv.). The resultant mixture was heated at reflux for 20 h then cooled to RT and concentrated *in vacuo*. The resultant residue was taken up in CH₂Cl₂ (10 mL + 3 x 2 mL washings) and filtered through sinter to partially remove the excess NaI. The filtrate was concentrated *in vacuo* and flash chromatography (20:80 \rightarrow 70:30 MeCN/EtOAc) afforded phosphonium salt 14 which contained trace amounts of NaI. The residue was taken up in CH₂Cl₂ (10 mL + 3 x 2 mL washings) and filtered to remove any remaining NaI. The filtrate was concentrated *in vacuo* to give a glassy solid. Lyophilisation with C₆H₆ (3x) afforded phosphonium salt 14 (38.9 mg, 94%) as a pale yellow powder: **R**_f 0.67 (30:70 EtOAc/MeCN); [α]²⁰ +9.3 (*c* 0.94, CHCl₃); **IR** (liquid film) 2932, 1612, 1585, 1513 cm⁻¹; ¹H NMR δ (400 MHz, C₆D₆) 7.73–7.84 (9H, m, ArH), 7.57 (2H, d, *J* = 8.5 Hz, ArH), 7.25 (2H, d, *J* = 8.6 Hz, ArH), 6.98–7.08 (6H, m, Ar<u>H</u>), 6.94

¹ D. A. Perrin and W. L. F. Armarego, *Purification of Laboratory Chemicals*, Pergamon Press, Oxford, 1988.

² I. Paterson, M. J. Coster, D. Y.-K. Chen, J. L. Aceña, J. Bach, L. E. Keown, and T. Trieselmann, Org. Biomol. Chem., 2005, X, XX-XX.

(2H, J = 8.5 Hz, ArH), 6.85 (2H, d, J = 8.5 Hz, ArH), 6.79 (2H, d, J = 8.6 Hz, ArH), 5.13 (1H, d, J)= 11.2 Hz, OCH_aH_bAr), 5.01 (2H, br s, 45-C=CH₂), 4.90–4.94 (2H, m, OCH_aH_bAr + OCH_aH_bAr), 4.88 (1H, d, J = 11.2 Hz, OCH_aH_bAr), 4.61–4.65 (1H, m, 29-CH_aH_b), 4.61 (1H, d, J = 10.8 Hz, $OCH_{a}H_{b}Ar$), 4.53 (1H, d, J = 11.2 Hz, $OCH_{a}H_{b}Ar$), 4.32 (1H, m, 33-CH), 4.16–4.23 (1H, m, 29- $CH_{a}H_{b}$), 4.03 (1H, s, 38-CH), 3.95 (1H, br d, J = 1.8 Hz, 35-CH), 3.54 (1H, br t, J = 8.8 Hz, 43-CH), 3.44 (3H, s, OCH₃), 3.43 (1H, m, 39-CH), 3.38 (3H, s, OCH₃), 3.34 (3H, s, OCH₃), 3.32 (1H, m, 42-CH), 3.30 (3H, s, OCH₃), 3.16 (1H, t, J = 9.6, 41-CH), 2.69 (1H, d, J = 14.9 Hz, 44-CH_aH_b), 2.61 (1H, dd, J = 15.4, 3.6 Hz, 36-CH_aH_b), 2.39–2.45 (2H, m, 30-CH_aH_b + 44-CH_aH_b), 2.23 (1H, m, 40-CH), 2.06 (1H, br d, J = 15.4 Hz, 36-CH_aH_b), 1.94 (3H, s, 46-CH₃), 1.90 (1H, m, 30-CH_aH_b), 1.74 (1H, m, 32-CH_aH_b), 1.56–1.65 (3H, m, 31-CH₂ + 34-CH), 1.21 (1H, m, 32-CH_aH_b), 1.08 (9H, s, SiC(CH₃)₃), 1.05 (3H, d, J = 6.2 Hz, 40-CHCH₃), 1.01 (3H, d, J = 7.1 Hz, 34-CHCH₃), 0.25 (3H, s, SiC<u>H</u>₃), 0.15 (3H, s, SiC<u>H</u>₃); ¹³C NMR δ (62.5 MHz, CDCl₃) 159.3, 159.2, 142.8, 135.2 (C parato P⁺), 133.7 (d, ${}^{3}J_{P-C} = 9.9$ Hz, C meta- to P⁺), 130.8, 130.7, 130.6 (d, ${}^{2}J_{P-C} = 12.4$ Hz, C ortho- to P^+), 129.5, 129.3, 118.1 (d, ${}^{1}J_{P-C} = 85.4$ Hz, C *ipso-* to P^+), 113.8, 113.8, 113.8, 112.5, 102.5, 86.9, 82.8, 79.6, 74.8, 74.6, 74.4, 73.4, 70.3, 66.3, 55.4, 55.3, 55.3, 47.5, 40.0, 38.5, 38.2, 32.9, 30.6, 27.3, 27.0, 25.8, 23.9, 223.0, 22.7, 17.9, 11.6 (d, ${}^{1}J_{P-C} = 152.4$ Hz, 29-<u>C</u>), -4.7, -4.7; m/z (+ESI) $1301 ([M + Na]^+, 42), 1151 (67), 587 (100).$

Side-chain truncated Wittig product (15)

CaH₂ (spatula tip) was added to phosphonium salt 14 (18.3 mg, 0.014 mmol) in a glove box and the system was evacuated and recharged with Ar. O₂-free THF (200 µL) was added, followed by HMPA (20 µL) and the mixture was stirred at RT for 2 h. The phosphonium salt/CaH₂ suspension was cooled to -78 °C and a freshly prepared solution of LiHMDS (0.253 M in THF, 90 µL, 0.023 mmol, 1.6 equiv.) was added, upon which the solution developed an intense yellow-orange colour. The reaction mixture was stirred at -78 °C for further 10 min. A 500 µL gas-tight syringe was flushed with a solution of aldehyde 2 (20.1 mg, 0.018 mmol, 1.3 equiv.) in THF (250 mL) and then the aldehyde solution was added to the vlide solution *via* this syringe. The resultant mixture was removed from the cooling bath and allowed to warm to RT. The mixture was stirred at RT for 40 min by which time a deep red colouration had developed. The reaction was quenched by transferring it, by pipette, to a cold (0 °C), vigorously stirred mixture of sat. aq. NH₄Cl/20% aq. Na₂S₂O₃ (2:1, 10 mL), and washing the flask with Et₂O (3 x 5 mL). The layers were separated and the aqueous phase was extracted with Et₂O (3 x 20 mL), the combined organic extracts were dried (Na₂SO₄) and concentrated *in vacuo*. Flash chromatography (10:90 \rightarrow 60:40 EtOAc/hexanes) afforded the desired Wittig product 15 (20 mg, 71%) as a white amorphous solid: $R_f 0.54$ (50:50 EtOAc/hexanes); [\alpha]²⁰ -13.0 (c 2.69, CDCl₃); IR (liquid film) 3416, 2953, 1734, 1612, 1513, 1461, 1371, 1249 cm⁻¹, ¹H NMR δ (500 MHz, C₆D₆) 7.44 (2H, d, J = 8.0 Hz, ArH), 7.35 (2H, d, J = 8.3 Hz, ArH), 7.25 (2H, d, J = 8.2 Hz, ArH), 6.81–6.85 (4H, m, ArH), 6.79 (2H, d, J = 8.3 Hz, ArH), 5.75 (1H, dd, J = 10.9, 7.1 Hz, 28-CH), 5.61–5.65 (2H, m, 15-CH + 27-CH), 5.55 (1H, m, 29-CH), 5.22 (1H, s, 13-C=CH_aH_b), 4.94–5.06 (6H, m, 5-CH + 13-C=CH_aH_b + 45-C=CH₂ + 2 x $OCH_{a}H_{b}Ar$), 4.87 (1H, d, J = 10.6 Hz, $OCH_{a}H_{b}Ar$), 4.77 (1H, d, J = 12.0 Hz, $OCH_{a}H_{b}CCl_{3}$), 4.75 $(1H, m, OCH_aH_bAr), 4.62 (1H, d, J = 10.5 Hz, OCH_aH_bAr), 4.55 (1H, d, J = 10.6 Hz, OCH_aH_bAr),$ 4.49 (1H, m, 3-CH), 4.47 (1H, d, J = 12.0 Hz, OCH₂H_bCCl₃), 4.33–4.36 (2H, m, 11-CH + 33-CH), 4.19 (1H, m, 19-CH), 3.91-3.94 (2H, m, 25-CH + 35-CH), 3.71 (1H, s, 38-CH), 3.54 (1H, m, 43-C<u>H</u>), 3.32 (6H, s, 2 x ArOC<u>H</u>₃), 3.30–3.34 (2H, m, 39-C<u>H</u> + 42-C<u>H</u>), 3.29 (3H, s, ArOC<u>H</u>₃), 3.25 (1H, m, 21-CH), 3.23 (3H, s, 37-COCH₃), 3.12-3.20 (1H, m, 41-CH), 3.06 (3H, s, 21-COCH₃), 3.05 (1H, m, 18-CH_aH_b), 2.92 (1H, m, 16-CH), 2.74–2.81 (2H, m), 2.58–2.73 (2H, m), 2.48 (2H, app d, J = 5.5 Hz, 12-CH₂), 2.34–2.45 (4H, m), 2.28–2.34 (2H, m), 2.21 (1H, m, 40-CH), 2.00–2.15 (3H, m), 1.95 (3H, s, COCH₃), 1.94 (3H, s, 46-CCH₃), 1.92 (1H, m, 6-CH_aH_b), 1.79 (3H, s, $COCH_3$, 1.71–1.80 (4H, m), 1.49–1.68 (6H, m), 1.42 (1H, d, J = 14.2 Hz, 10- CH_aH_b), 1.23 (3H, d, J = 6.8 Hz, 14-CHCH₃), 1.22–1.24 (1H, m, 6-CH_aH_b), 1.18 (3H, d, J = 6.9 Hz, 16-CHCH₃), 1.17 (1H, m, 4-CH_aH_b), 1.12 (9H, s, SiC(CH₃)₃), 1.11–1.14 (1H, m, 24-CH_aH_b), 1.03–1.06 (1H, m, 20CH_a<u>H</u>_b), 1.03 (9H, t, J = 7.0 Hz, Si(CH₂C<u>H₃</u>)₃), 1.02 (9H, s, SiC(C<u>H₃</u>)₃), 0.99 (3H, s, 9-CC<u>H₃</u>), 0.97–1.03 (5H, m, 8-CH_a<u>H</u>_b + 10-CH_a<u>H</u>_b + 34-CHC<u>H₃</u>), 0.82 (3H, d, J = 6.0 Hz, 40-CHC<u>H₃</u>), 0.57–0.60 (6H, m, Si(C<u>H₂</u>CH₃)₃), 0.24 (3H, s, SiC<u>H₃</u>), 0.14 (3H, s, SiC<u>H₃</u>), 0.08 (6H, s, 2 x SiC<u>H₃</u>); ¹³C **NMR** δ (62.5 MHz, C₆D₆) 209.3, 169.9, 168.9, 159.6, 159.3, 147.7, 143.1, 132.0, 131.5, 131.3, 130.7, 130.4, 129.2, 129.0, 113.8, 113.8, 113.8, 113.5, 112.8, 102.8, 98.1, 96.9, 95.3, 86.9, 83.1, 79.7, 77.9, 76.2, 74.6, 74.4, 74.2, 73.8, 73.6, 70.9, 70.4, 67.3, 66.6, 66.4, 65.0, 64.3, 62.0, 60.6, 54.9, 54.5, 54.5, 54.5, 50.9, 47.7, 47.6, 47.0, 45.2, 44.3, 42.7, 40.6, 40.1, 39.2, 38.9, 38.4, 37.7, 34.7, 33.9, 33.0, 31.9, 31.3, 30.2, 29.9, 28.5, 26.6, 25.9, 23.0, 21.1, 20.3, 18.1, 13.4, 13.2, 12.0, 10.3, 7.3, 6.9, -4.6, -4.7, -5.0, -5.0; **HRMS** (+ESI) Calc. for C₁₀₃H₁₆₁O₂₄Cl₃Si₃Na [M + Na]⁺: 1993.9643, found: 1993.9435.

Side-chain truncated analogue - tris-PMB deprotection (16)

To a cooled (0 °C) solution of *tris*-PMB ether 15 (9.4 mg, 4.76 µmol) in CH₂Cl₂/pH7 buffer (10:1, 1.1 mL) was added DDQ (97 mg, 0.43 mmol, 90 equiv.) over a period of 1.5 h (10 equiv. every 10 min). The reaction was quenched by pouring into sat. aq. NaHCO₃ (10 mL) and the layers were separated. The aqueous phase was extracted with Et₂O (3 x 15 mL), combined organics were dried (Na₂SO₄) and concentrated *in vacuo*. Flash chromatography (20:80 \rightarrow 100:0 EtOAc/hexanes) afforded the desired *tris*-PMB deprotected material 16 as an inseparable mixture (5.3 mg, 69%) as a white amorphous solid consisting of an inseparable ca. 1.1:1 mixture of methyl and hemiacetal at C37: \mathbf{R}_{f} 0.50 (80:20 EtOAc/hexanes); IR (liquid film) 3449, 2954, 1735 (C=O), 1654 cm⁻¹; ¹H **NMR** δ (500 MHz, C₆D₆) 5.77 (1H, m, 28-CH), 5.71 (1H, br d, J = 9.7 Hz, 15-CH), 5.67 (1H, m, 27-CH), 5.59 (1H, m, 29-CH), 5.31 (1H, s, 13-C=CH_aH_b), 5.22 (0.5H, s, 37-OH), 5.08 (1H, s, 13-C=CH_aH_b), 5.06 (1H, br s, 5-CH), 5.01 (1H, s, 45-C=CH_aH_b), 4.98 (1H, s, 45-C=CH_aH_b), 4.87 (1H, d, J = 11.9 Hz, $CH_{a}H_{b}CCl_{3}$), 4.58 (1H, d, J = 12.0 Hz, $CH_{a}H_{b}CCl_{3}$), 4.58 (1H, m, 3-CH), 4.43 (2H, m, 33-CH + 11-CH), 4.27 (1H, m, 19-CH), 3.98 (2H, m, 25-CH + 35-CH), 3.89 (0.5H, d, J = 6.2 Hz), 3.86 (0.5H, d, *J* = 10.3 Hz), 3.72 (0.5H, d, *J* = 8.6 Hz), 3.54 (0.5H, m, 43-C<u>H</u>), 3.31–3.46 (2H, m), 3.26 (1.5 H, s, 37-OCH₃), 3.01–3.22 (4H, m), 3.16 (3H, 2 x s, 21-OCH₃), 2.84–2.93 (2H, m), 2.65–2.76 (3H, m), 2.56 (2H, d, J = 6.2 Hz), 2.29–2.46 (6H, m), 2.20 (4H, m), 2.05 (3H, 2 x s, COCH₃), 2.04 (1H, m), 1.92 (3H, 2 x s, COCH₃), 1.92 (3H, s, 46-CCH₃), 1.80–1.92 (4H, m), 1.61– 1.74 (6H, m), 1.47–1.53 (4H, m), 1.33 (3H, m, 14-CHCH₃), 1.29 (3H, m, 16-CHCH₃), 1.22–1.34 (2H, m), 1.06–1.18 (33H, m), 0.95 (3H, m, 40-CHCH₃), 0.68 (6H, q, J = 7.6 Hz, Si $(CH_2CH_3)_3$), 0.13–0.29 (12H, m, 3 x Si(CH₃)₂); m/z (+ESI) 1635 (for 16Me, [M + Na]⁺, 43), 1621 (for 16H, [M $+ \text{Na}^{+}$, 83), 1151 (18), 822 (100).

Side-chain truncated analogue – TCE deprotection (17)

To a solution of **16** (36.8 mg, max. 0.023 mmol) in THF (1.0 mL) was added Zn dust (1.0 g, excess). The resultant suspension was stirred at RT for 2 min and aq. NH₄OAc (1M, 100 μ L) was added. The reaction mixture was stirred at RT for 30 min and then diluted with EtOAc (10 mL). The supernatant was separated with a pipette and the resultant Zn residue was washed with EtOAc (3 x 10 mL). The combined organics were dried (Na₂SO₄), filtered through sinter, and concentrated *in vacuo*. Flash column chromatography (40:60 \rightarrow 100:0 EtOAc/hexanes, then 1:99 AcOH/EtOAc) afforded *seco*-acid **17** (20.9 mg, 62%), as a mixture of methyl acetal and hemiacetal at C37, and partially dechlorinated esters (5.7 mg, 16%) as an amorphous solid and a colourless oil, respectively.

Seco-acid 17: $\mathbf{R}_{\mathbf{f}} 0.36$ (80:20 EtOAc/hexanes); **IR** (liquid film) 3458, 2932, 1732 (C=O), 1372 cm⁻¹; ¹H NMR δ (500 MHz, C₆D₆) 5.79 (2H, m, 15-C<u>H</u> +28-C<u>H</u>), 5.66 (1H, m, 27-C<u>H</u>), 5.57 (1H, m, 29-C<u>H</u>), 5.04–5.30 (5.5H, m, 37-O<u>H</u> + 5-C<u>H</u> + 13-C=C<u>H</u>₂, 45-C=C<u>H</u>₂), 4.62, 4.53, 4.29–4.40 (4H, 3 x m, 3-C<u>H</u> + 11-C<u>H</u> + 19-C<u>H</u> + 33-C<u>H</u>), 3.81–4.00 (3H, m, 25-C<u>H</u> + 35-C<u>H</u> + 38-C<u>H</u> + 39-C<u>H</u>), 3.23–3.57 (3H, m), 3.31 (1.5H, s, 37-OC<u>H</u>₃), 2.97–3.19 (3H, m), 3.14, 3.15 (3H, 2 x s, 21-OC<u>H</u>₃), 2.75–2.94 (3H, m), 2.67 (2H, m), 2.43–2.56 (4H, m), 2.25–2.42 (3H, m), 2.19 (4H, m), 2.04 (3H, s, COC<u>H</u>₃), 2.02 (1H, m), 1.98, 1.93 (3H, 2 x s, COC<u>H</u>₃), 1.95 (3H, s, 46-C<u>H</u>₃), 1.75–1.85 (4H, m),

1.68 (6H, m), 1.02–1.52 (46H, m), 0.73 (6H, m, Si(CH₂CH₃)₃), 0.10–0.29 (12H, m, 3 x Si(CH₃)₂). The absence of doublet peaks at 4.58 and 4.87 indicated the loss of the 2,2,2-trichloroethyl ester moiety; $\mathbf{m/z}$ (+ESI) 1503 (for **17Me**, [M + Na]⁺, 34), 1489 (for **17H**, [M + Na]⁺, 67), 756 (56), 681 (94), 469 (100).

Side-chain truncated analogue – macrolactonisation (S1)



To a solution of seco-acid 17 (20.9 mg, max. 0.014 mmol) in THF (200 µL) was added freshly prepared solution of Et₃N (0.5M in THF, 141 µL, 0.07 mol, 5 equiv.) and a solution of 2,4,6trichlorobenzovl chloride (0.5M in THF, 85 µL, 0.042 mmol, 3 equiv.). The resultant solution was left to stir at RT for 1 h before a further aliquot of reagents (Et₃N and 2,4,6-trichlorobenzoyl chloride, same stoichiometry as above) were added. This process was repeated for a third time such that a total of Et₃N (15 equiv.) and 2,4,6-trichlorobenzoyl chloride (9 equiv.) were added in three portions over a period of 3 h. The reaction mixture was diluted with PhMe (1 mL) and this solution was then added to a refluxing solution of DMAP (34 mg, 0.28 mmol, 20 equiv.) in PhMe (10 mL) over a period of 3 minutes. The anhydride flask was rinsed with PhMe (2 x 1 mL) and added to the refluxing reaction. The resultant cloudy white suspension was heated at reflux for 16 h and then cooled to RT before being quenched with sat. NaHCO₃ (20 mL). The layers were separated and the aqueous phase was extracted with EtOAc (3 x 20 mL). The combined organic extracts were washed with brine (20 mL), dried (Na₂SO₄) and concentrated *in vacuo*. Flash chromatography (20:80 \rightarrow 80:20 EtOAc/hexanes) afforded macrolactone S1 (9.7 mg, 47%) as a colourless oil, consisting of an inseparable mixture of methyl acetal and hemiacetal at C37: Rf 0.32 (50:50 EtOAc/hexanes); IR (liquid film) 3455, 2953, 1736 (C=O), 1459 cm⁻¹; ¹H NMR δ (500MHz, CD₃CN) 5.42 (1H, m, 28-CH), 5.36 (1H, m, 29-CH), 5.18 (1H, m, 27-CH), 5.11 (1H, dd, J = 10.5, 1.5 Hz, 15-CH), 4.91 (1H, br s, 5-CH), 4.85 (2H, br s, 13-C=CH₂), 4.78–4.83 (2H, m, 45-C=CH₂), 4.73, 4.68 (1H, m, dd, J = 10.6, 9.1 Hz, 41-CH), 4.29 (2H, m, 3-CH + 11-CH), 4.13 (2H, m), 4.04 (1H, m), 3.89 (1.5H, m), 3.80 (0.5H, m), 3.60 (1H, d, J = 10.3 Hz, 39-CH), 3.49 (1H, m, 21-CH), 3.39 (2H, m, 38-CH + 43-CH), 3.31 (1H, d, J = 6.1 Hz), 3.25, 3.24 (3H, 2 x s, 21-OCH₃), 3.03–3.21 (3H, m), 2.92 (1H, m), 2.84 (1H, d, J = 6.3 Hz), 2.67–2.80 (3H, m), 2.36–2.55 (3H, m), 2.31 (2H, m), 2.04–2.18 (7H, m), 1.95 (3H, s, COCH₃), 1.87 (3H, s, COCH₃), 1.87 (2H, m), 1.75 (2H, m), 1.55–1.66 (6H, m), 1.32– 1.48 (8H, m), 1.21 (3H, s, 9-CC<u>H</u>₃), 1.19 (3H, d, *J* = 7.0 Hz, 16-CHC<u>H</u>₃), 1.14 (1H, m), 1.08 (2H, m), 1.08 (3H, d, J = 6.9 Hz, 14-CHCH₃), 0.84–0.97 (33H, m), 0.59 (6H, m, Si(CH₂CH₃)₃), 0.02– 0.12 (12H, m, 3 x Si(CH₃)₂). The appearance of double-doublet signal at δ 4.68 (J = 10.6, 9.1 Hz) indicated the formation of the macrolactones; m/z (+ESI) 1485 (for S1Me, $[M + Na]^+$, 19), 1471 (for S1H, $[M + Na]^+$, 93), 747 (100).

Side-chain truncated analogue (18)

To a cold (0 °C) solution of macrolactone **S1** (2.4 mg, max. 1.64 μ mol) in MeCN (0.5 mL) was added a freshly prepared HF solution (40% HF_(aq)/H₂O/MeCN, 0.5/0.9/8.6, 50 μ L) and the reaction mixture was stirred at 0 °C for 4 h before being quenched with sat. aq. NaHCO₃ (5 mL). The

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resultant mixture was diluted with CH₂Cl₂ (5 mL) and the layers were separated. The aqueous phase was extracted with CH₂Cl₂ (3 x 10 mL), combined organics were dried (Na₂SO₄) and concentrated in vacuo. Flash chromatography (5:95 \rightarrow 10:90 MeOH/CH₂Cl₂) afforded 18 as a white solid: R_f 0.24 (5:95 MeOH/CH₂Cl₂); ¹H NMR δ (500 MHz, CD₃CN) 5.47 (1H, m, 29-CH), 5.36 (1H, br t, J = 9.7 Hz, 28-CH), 5.18 (1H, dd, J = 10.6, 1.8 Hz, 15-CH), 5.00 (1H, m, 27-CH), 4.94 (1H, br s, 5-CH), 4.85 (2H, br s, 13-C=CH₂), 4.84 (1H, br s, 45-C=CH_aH_b), 4.77 (1H, dd, J = 11.0, 9.2 Hz, 41-CH), 4.74 (2H, br s, 45-C=CH_aH_b + 37-OH), 4.35 (1H, br t, J = 11.0 Hz, 11-CH), 4.29 (1H, d, J =10.7 Hz, 25-O<u>H</u>), 4.26 (1H, br s, 9-O<u>H</u>), 4.26 (1H, obsc., 3-C<u>H</u>), 4.19 (1H, br d, J = 5.2 Hz, 42-OH), 4.16 (1H, m, 33-CH), 4.02 (1H, br t, J = 11.1 Hz, 19-CH), 3.92 (1H, m, 25-CH), 3.82 (1H, d, J = 9.3 Hz, 35-OH), 3.74 (1H, br d, J = 10.4 Hz, 39-CH), 3.66 (1H, m, 35-CH), 3.49 (1H, m, 21-CH), 3.38 (1H, obsc., 43-CH), 3.37 (1H, d, J = 10.7 Hz, 38-CH), 3.27 (3H, s, 21-OCH₃), 3.14 (1H, td, J = 9.1, 5.2 Hz, 42-CH), 3.04 (1H, dq, J = 10.7, 6.9 Hz, 16-CH), 2.82–2.90 (3H, m, 14-CH + 18- $CH_{a}H_{b} + 38-OH$, 2.61–2.65 (2H, m, 18- $CH_{a}H_{b} + 44-CH_{a}H_{b}$), 2.56 (1H, dd, J = 16.7, 1.8 Hz, 2- CH_aH_b), 2.49 (1H, dd, J = 16.7, 10.5 Hz, 2- CH_aH_b), 2.29 (2H, m, 24- $CH_aH_b + 12-CH_aH_b$), 2.14 (3H, obsc., 45-CCH₃), 2.04–2.10 (4H, m, 12-CH_aH_b + 30-CH₂ + 44-CH_aH_b), 1.90–1.99 (3H, m, 20-CH_aH_b + 22-CH_aH_b + 36-CH_aH_b), 1.85 (3H, s, COCH₃), 1.73 (3H, s, COCH₃), 1.68 (1H, m, 4- $CH_{a}H_{b}$), 1.54–1.62 (9H, m, 4- $CH_{a}H_{b}$ + 8- $CH_{a}H_{b}$ + 10- $CH_{a}H_{b}$ + 24- $CH_{a}H_{b}$ + 26- CH_{2} + 31- $CH_{a}H_{b}$ + $34-CH + 36-CH_{a}H_{b}$, 1.48 (1H, d, J = 14.0 Hz, 8-CH_aH_b), 1.41 (1H, m, 32-CH_aH_b), 1.27-1.32 (3H, m, $10-CH_{a}H_{b} + 31-CH_{a}H_{b} + 32-CH_{a}H_{b}$), $1.17 (3H, d, J = 7.0 Hz, 16-CHCH_{3})$, 1.11 (1H, t, J = 11.9)Hz, 22-CH_aH_b), 1.07 (3H, s, 9-CCH₃), 1.04 (3H, d, J = 6.9 Hz, 14-CHCH₃), 0.96 (1H, ddd, J = 12.0, 12.0, 12.0 Hz, 20-CH_aH_b), 0.85 (3H, d, J = 7.2 Hz, 34-CHCH₃), 0.77 (3H, d, J = 6.7 Hz, 40- $CHCH_3$).

Table S1: NMR assignments for natural and synthetic altohyrtin A/spongistatin 1 (1) recorded in CD₃CN. Coupling constants are in Hz (in parentheses)



¹ H/ ¹³ C No.	Natural	Synthetic	Natural	Synthetic
	¹³ C	¹³ C	$^{1}\mathrm{H}$	$^{1}\mathrm{H}$
	(100 MHz)	(125 MHz)	(400 MHz)	(500 MHz)
1	173.07	173.07		
2	40.86	40.86	2.44 dd (10, 18)	2.45 dd (10.5, 16.2)
			2.53 dd (2, 18)	2.52 dd (1.8, 16.2)
3	63.59	63.59	4.25 brt (10)	4.24 brt (10.7)
4	34.65	34.65	1.55*, 1.68*	1.55*, 1.68*
5	67.06	67.06	4.92 brs	4.92 brs
6	38.17	38.16	1.67 dd (5, 14); 1.78 brd (14)	1.66 dd (4.4, 15.2); 1.78 d (15.1)
7	99.26	99.25		
8	46.76	46.75	1.47 d (14), 1.60*	1.46 d (14.0), 1.60*
9	69.64	69.63		
9a	30.21	30.21	1.06 s	1.06 s
10	44.96	44.95	1.28*, 1.55*	1.28*, 1.55*
11	64.70‡	64.70	4.25 brt (10)	4.24 brt (10.7)
12	44.24	44.24	1.99*; 2.27 brd (14)	1.99*; 2.26 brd (14.0)
13	148.03	148.03		· · · · · · · · · · · · · · · · · · ·
13a	114.86	114.86	4.83 brs; 4.83 brs	4.83 brs; 4.83 brs
14	36.60	36.59	2.78*	2.78*
14a	12.09	12.09	1.04 d (6.9)	1.04 d (6.9)
15	75.34	75.34	5.12 dd (1.7, 11)	5.11 dd (1.8, 10.7)
16	47.62	47.61	3.04 dq (7, 11)	3.04 dq (6.9, 10.7)
16a	13.73	13.73	1.15 d (7)	1.15 d (6.9)
17	213.52	213.52		
18	51.94	51.95	2.62 brd (18); 2.86 dd (11, 18)	2.61 brd (18.2); 2.86 dd (10.1,
				19.4)
19	66.16	66.15	4.00 brt (11)	3.99 brt (11.1)
20	37.70	37.70	0.97 ddd (12, 12, 12); 1.98*	0.96 ddd (12, 12, 12); 1.98*
21	73.98	73.98	3.46 tt (4, 4, 12, 12)	3.46 tt (4.2, 4.2, 11.6, 11.6)
22	44.18	44.18	1.08 t (12); 1.99*	1.08 t (12.1), 1.99*
23	99.91	99.91		
24	34.91	34.90	1.55*; 2.28*	1.55*; 2.27*
25	64.41	64.41	3.93 brm	3.94 brm
26	39.11	39.11	1.57*; 1.57*	1.57*; 1.57*
27	61.22	61.21	5.00 ddd (4.3, 10, 10)	5.00 ddd (3.8, 9.9, 9.9)
28	131.22	131.21	5.32 brt (10)	5.32 brt (10.8)
29	133.42	133.41	5.48 ddd (10, 10, 10)	5.48 ddd (9.9, 9.9, 10.6)
30	28.07	28.08	2.00*; 2.19*	2.00*; 2.18*
31	27.04	27.04	1.23*; 1.60*	1.23*; 1.60*

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32	32.82	32.83	1.30 m; 1.42 m	1.30 m; 1.42 m			
33	67.15	67.15	4.13 dt (3.4, 3.4, 8)	4.13 dt (3.0, 3.0, 9.0)			
34	39.32	39.32	1.57 m	1.57 m			
34a	11.55	11.55	0.81 d (7)	0.81 d (7.2)			
35	71.47	71.47	3.65 brs	3.65 m			
36	33.79	33.78	1.61*; 1.89*	1.61*; 1.89*			
37	99.41	99.41					
38	73.11	73.09	3.34 brs	3.34 d (10.7)			
39	81.30	81.30	3.72 brd (10)	3.72 brd (10.2)			
40	37.26	37.26	1.91*	1.91*			
40a	12.69	12.69	0.74 d (7)	0.74 d (6.6)			
41	80.60	80.60	4.75 dd (9, 11)	4.75 dd (9.2, 11.0)			
42	73.11	73.09	3.12 t (9)	3.11 dt (5.4, 9.1)			
43	78.72	78.70	3.39 brt (9)	3.38 dt (2.2, 8.6)			
44	40.24	40.23	2.08*; 2.76 brd (13)	2.08*; 2.75 brd (13.9)			
45	144.00	143.98					
45a	116.61	116.63	4.86 brs; 4.89 brs	4.86 brs; 4.89 brs			
46	43.93	43.92	2.33 brdd (7, 14); 2.19*	2.33 brdd (6.3, 13.6); 2.18*			
47	70.13	70.12	4.36 ddd (6, 7, 11)	4.36*			
48	139.21	139.20	6.11 dd (6, 15)	6.11 dd (5.7, 15.0)			
49	126.99	126.98	6.41 brd (15)	6.41 dd (1.1, 15.0)			
50	139.21	139.20					
51	116.48	116.49	5.35 brs; 5.45 brs	5.35 brs; 5.45 brs			
OMe	55.72	55.71	3.24 s	3.24 s			
OAc	21.78	27.78	1.94 s	1.94 s			
	171.61	171.61					
OAc	21.00	21.00	1.84 s	1.84 s			
	170.21	170.19					
OH (C9)			4.32 brs	4.31 brs			
OH (C25)			4.39 d (9.9)	4.39 d (10.0)			
OH (C35)			3.83 brm	3.82 d (9.4)			
OH (C37)			4.73 d (2)	4.72 d (2.5)			
OH (C38)				2.86 brd (10.1)			
OH (C42)				4.35 brd (5.3)			
OH (C47)				3.51 brd (2.5)			

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* coupling constants for these signals were not measured due to overlapping. ‡ correction to Pettit's published value.

1.0 1.5 2.0 5.5 -3.0 V IMMI L 3.5 4.0 4.5 . 2.0 . 5.5 ¹H NMR of synthetic altohyrtin A / spongistatin 1 (1) (500 MHz, CD₃CN) 6.6 - 0.5 ppm . 9.9 - udd Ы ģ ō



















